



Flow CCS functionality

Alf B. Rustad with **huge** help from Tor Harald Sandve

CO2STORE in Flow

```

-- =====
--
-- RUNSPEC SECTION
--
-- =====
RUNSPEC
-----
-- FLUID TYPES AND TRACER OPTIONS
-- -----
--
--         ACTIVATE CO2 STORAGE IN THE MODEL (OPM FLOW CO2 STORAGE KEYWORD)
--
CO2STORE
--
--         ACTIVATE GAS-WATER THE MODEL (OPM FLOW KEYWORD)
--
GASWAT
--
--         DISSOLVED GAS IN WATER IS PRESENT IN THE RUN (OPM FLOW KEYWORD)
--
DISGASW
--
--         VAPORIZED WATER IN DRY/WET GAS IS PRESENT IN THE RUN (OPM FLOW KEYWORD)
--
VAPWAT
  
```

Salinity in Flow

The first example activates the standard Brine model and has no terminating “/”.

```
--  
--          ACTIVATE STANDARD BRINE MODEL IN THE RUN  
--  
BRINE
```

The second example illustrates how to activate OPM Flow’s Salt Precipitation model.

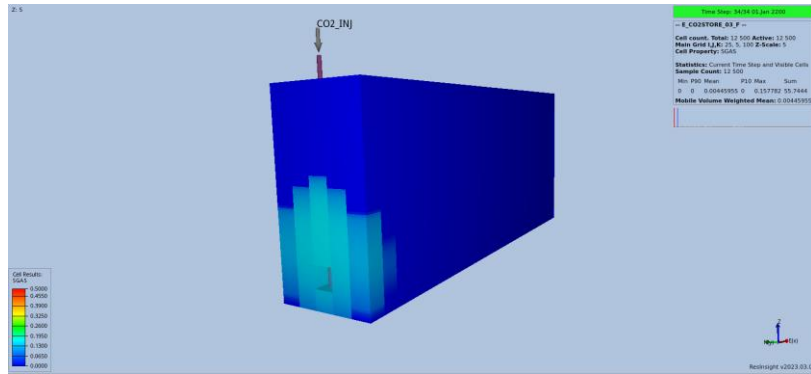
```
--  
--          ACTIVATE STANDARD BRINE MODEL IN THE RUN  
--  
BRINE  
--  
--          ACTIVATE THE OPM FLOW SALT PRECIPITATION MODEL (OPM FLOW KEYWORD)  
--  
PRECSALT  
--  
--          VAPORIZED WATER IN DRY/WET GAS IS PRESENT IN THE RUN (OPM FLOW KEYWORD)  
--  
VAPWAT
```

Case 1; synthetic models

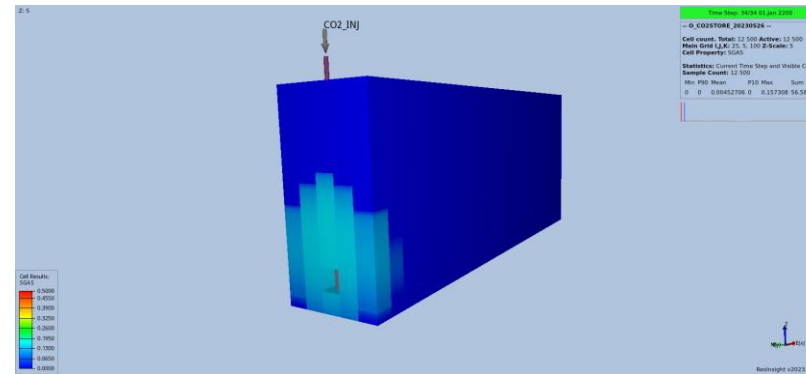
- The models are homogeneous and isothermal
- Injection rate is $1 \times 10^6 \text{ Sm}^3$ (0.68 mtpa) from a single injector in 25 years
- With and without salinity

The grids

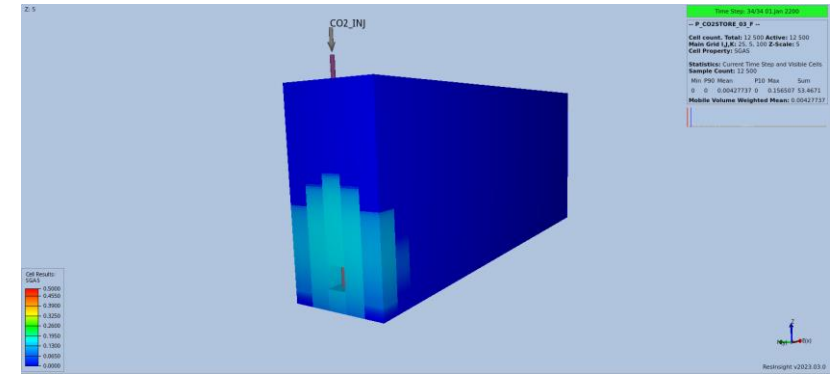
E300



OPM Flow

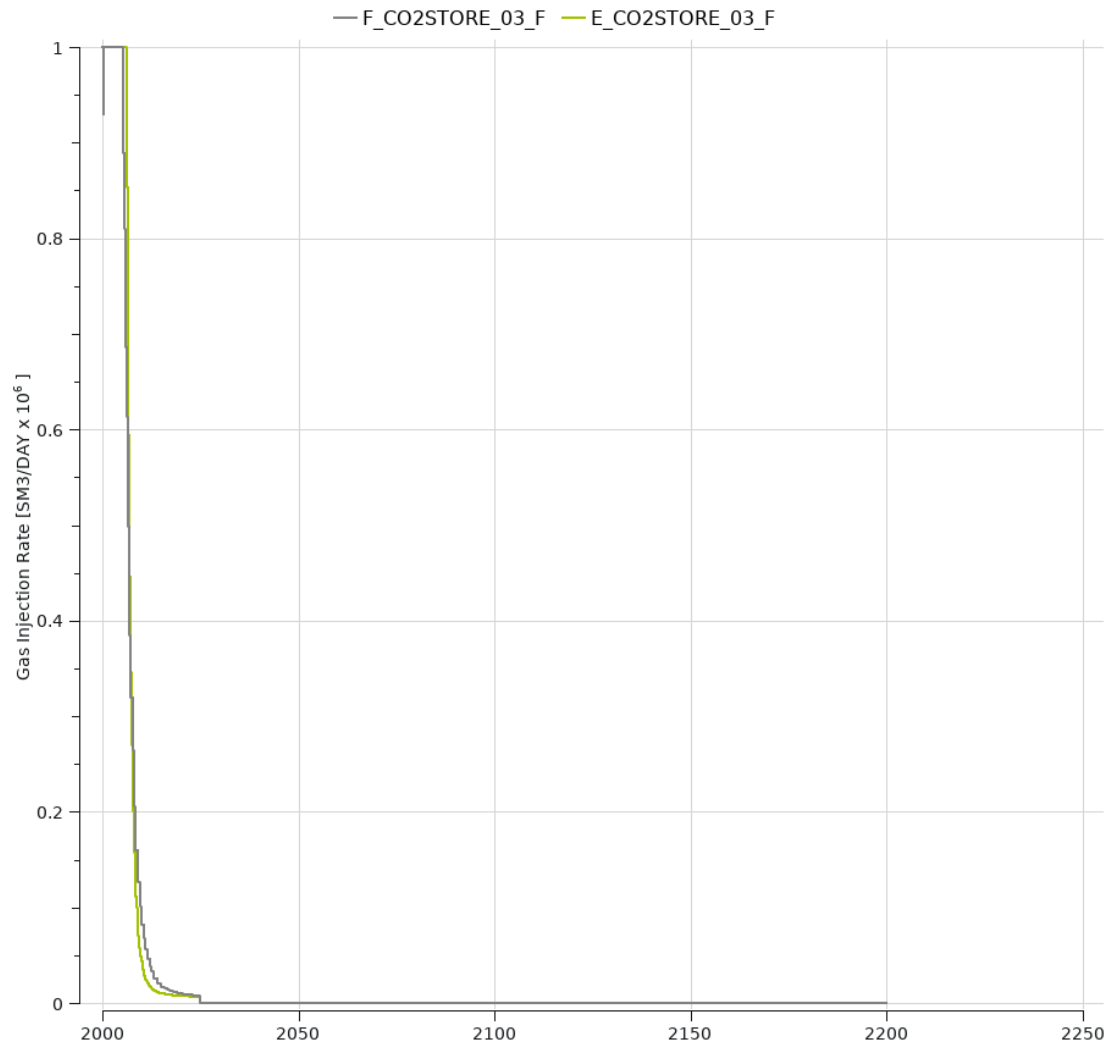


Pflotran

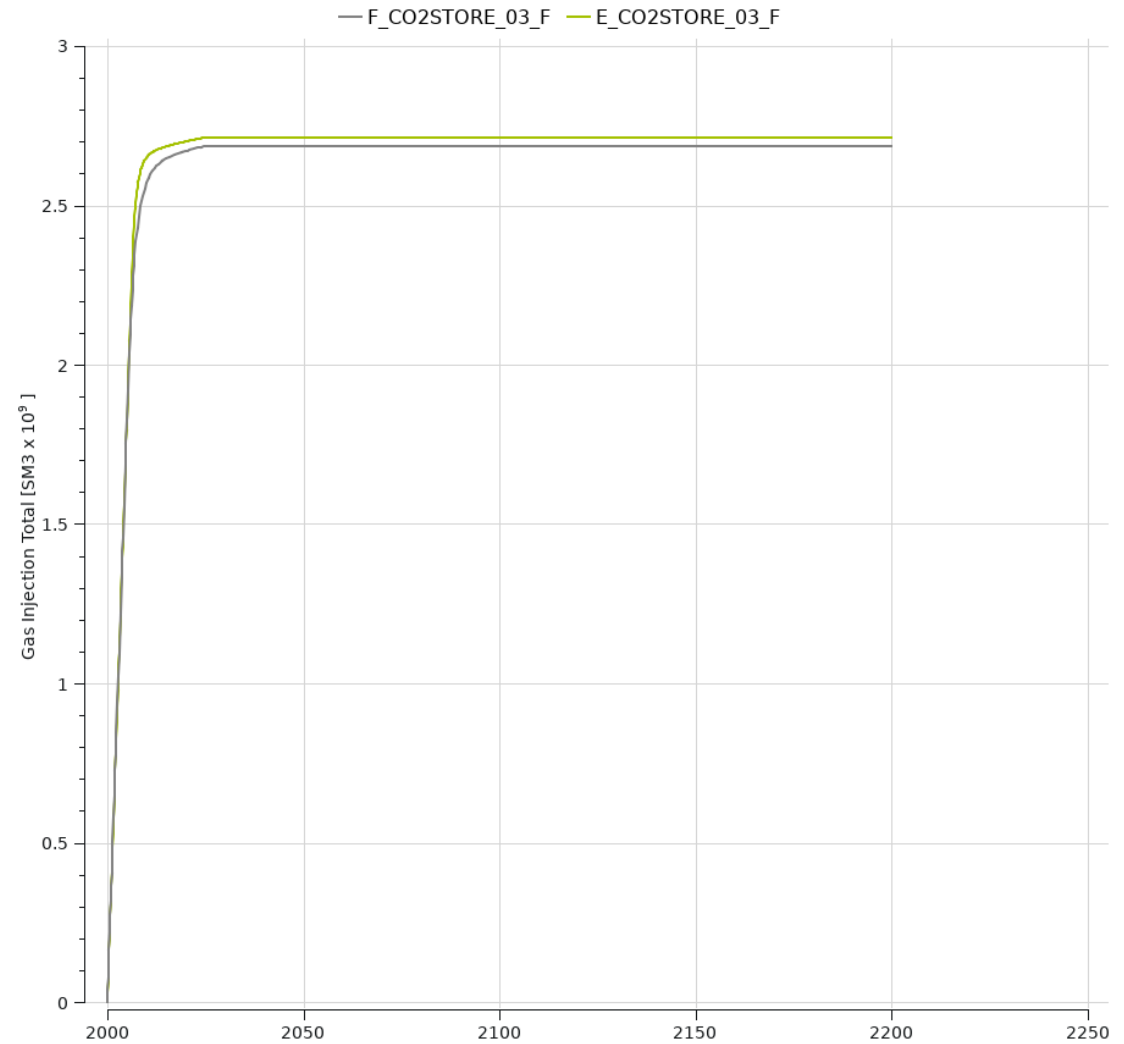


Gas injection

Gas Injection Rate



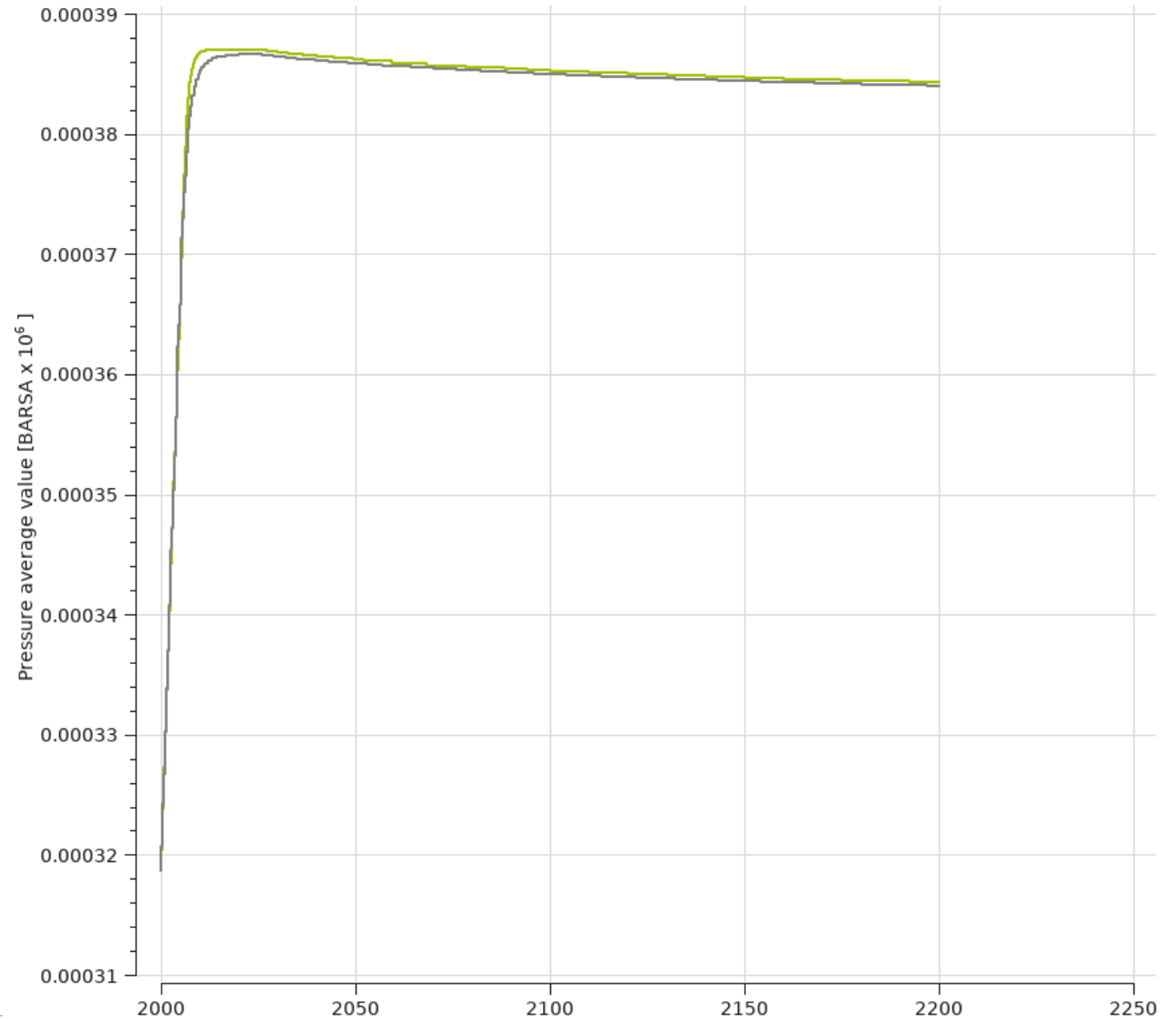
Gas Injection Total



Pressure average value

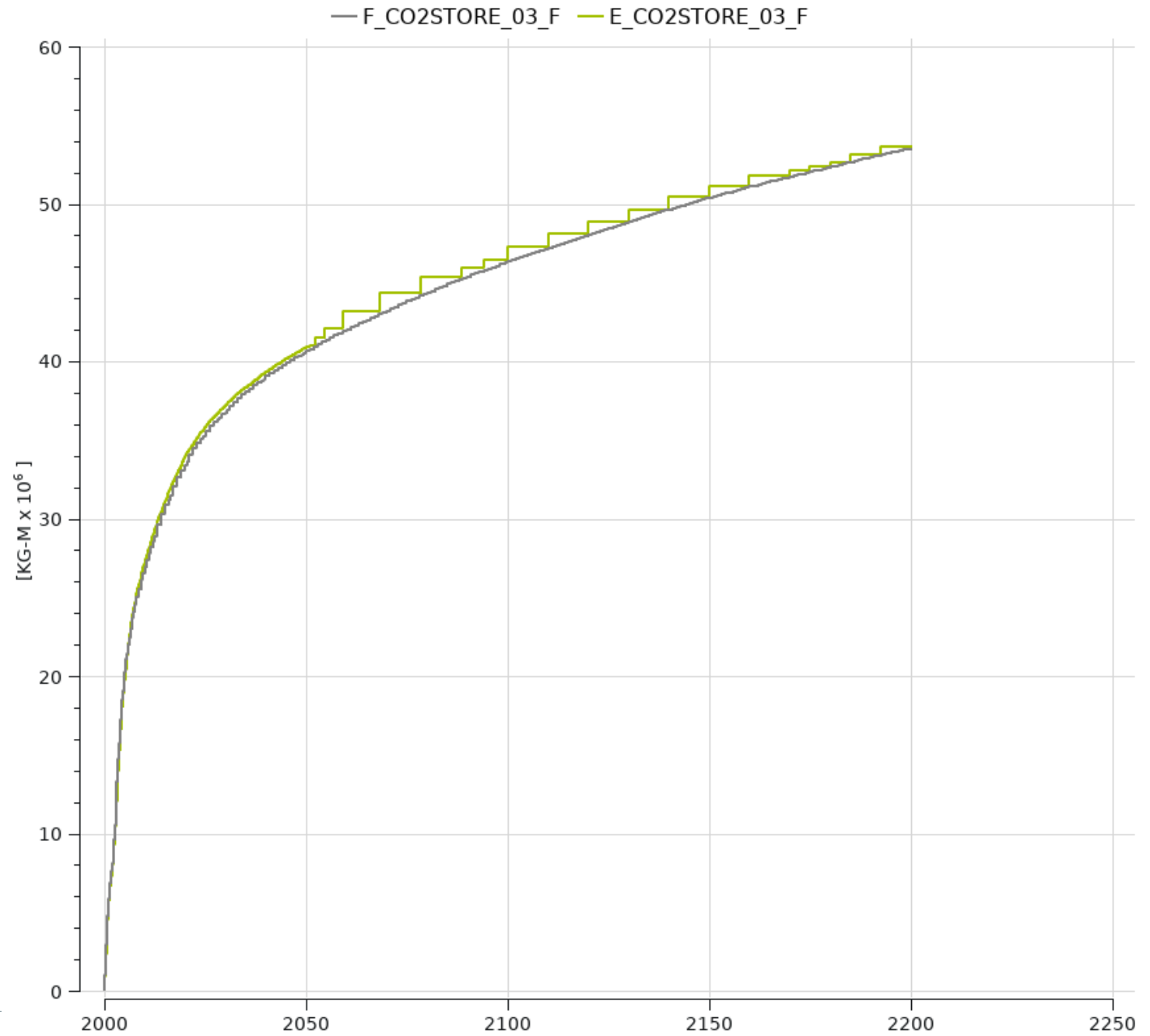
— E_CO2STORE_03_F — F_CO2STORE_03_F

Average reservoir pressure



Dissolved CO2 mass

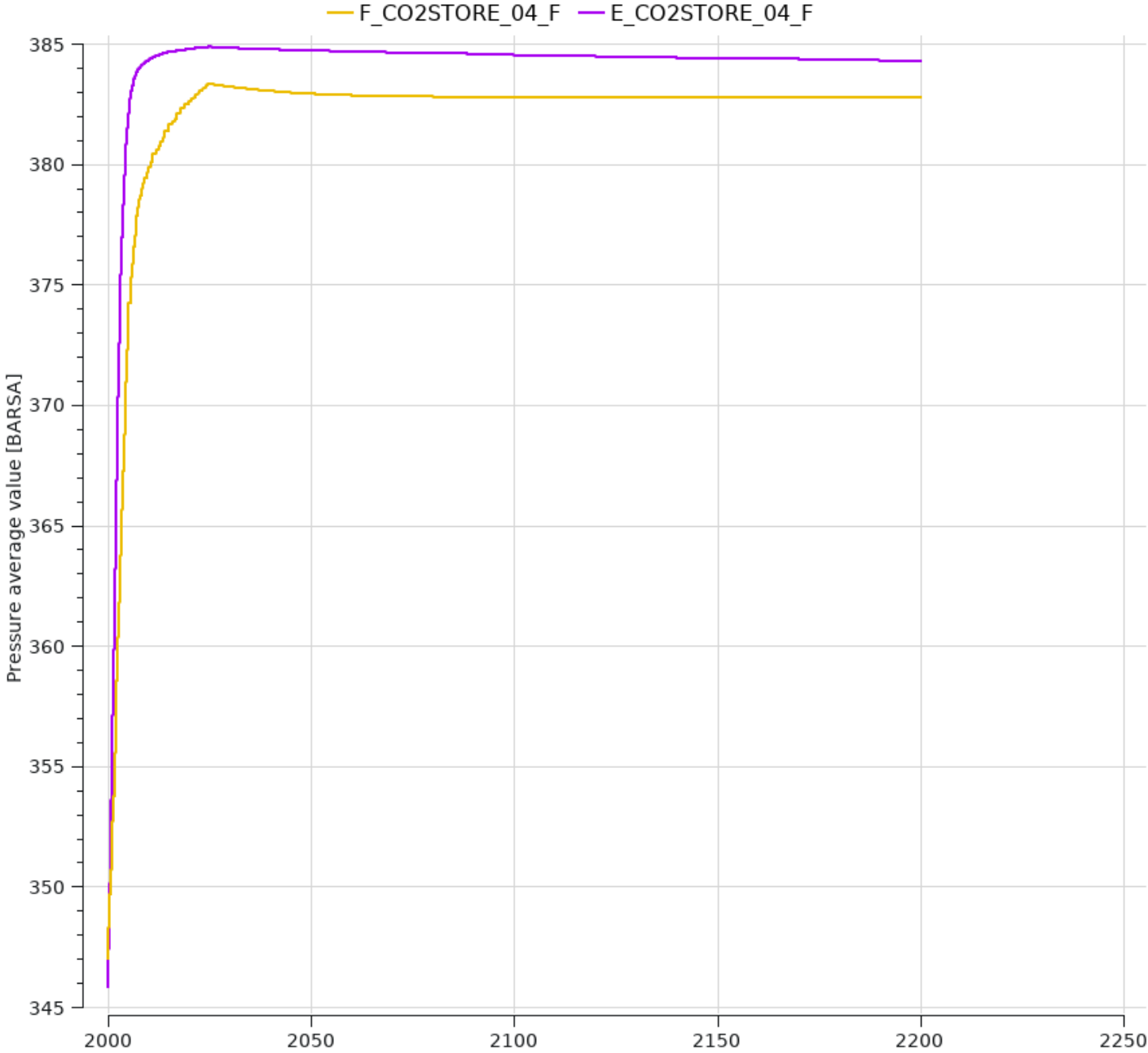
FWCD



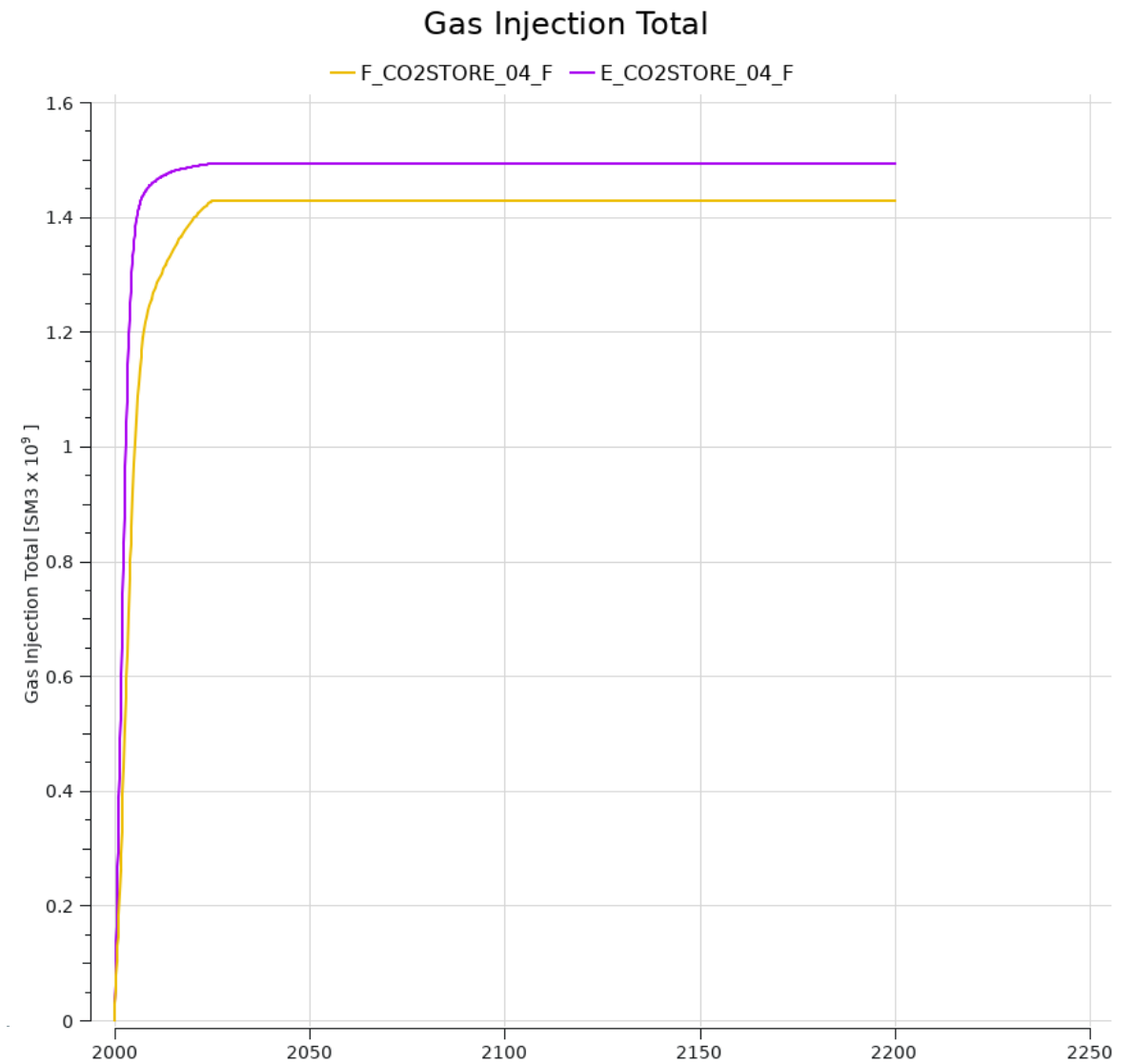
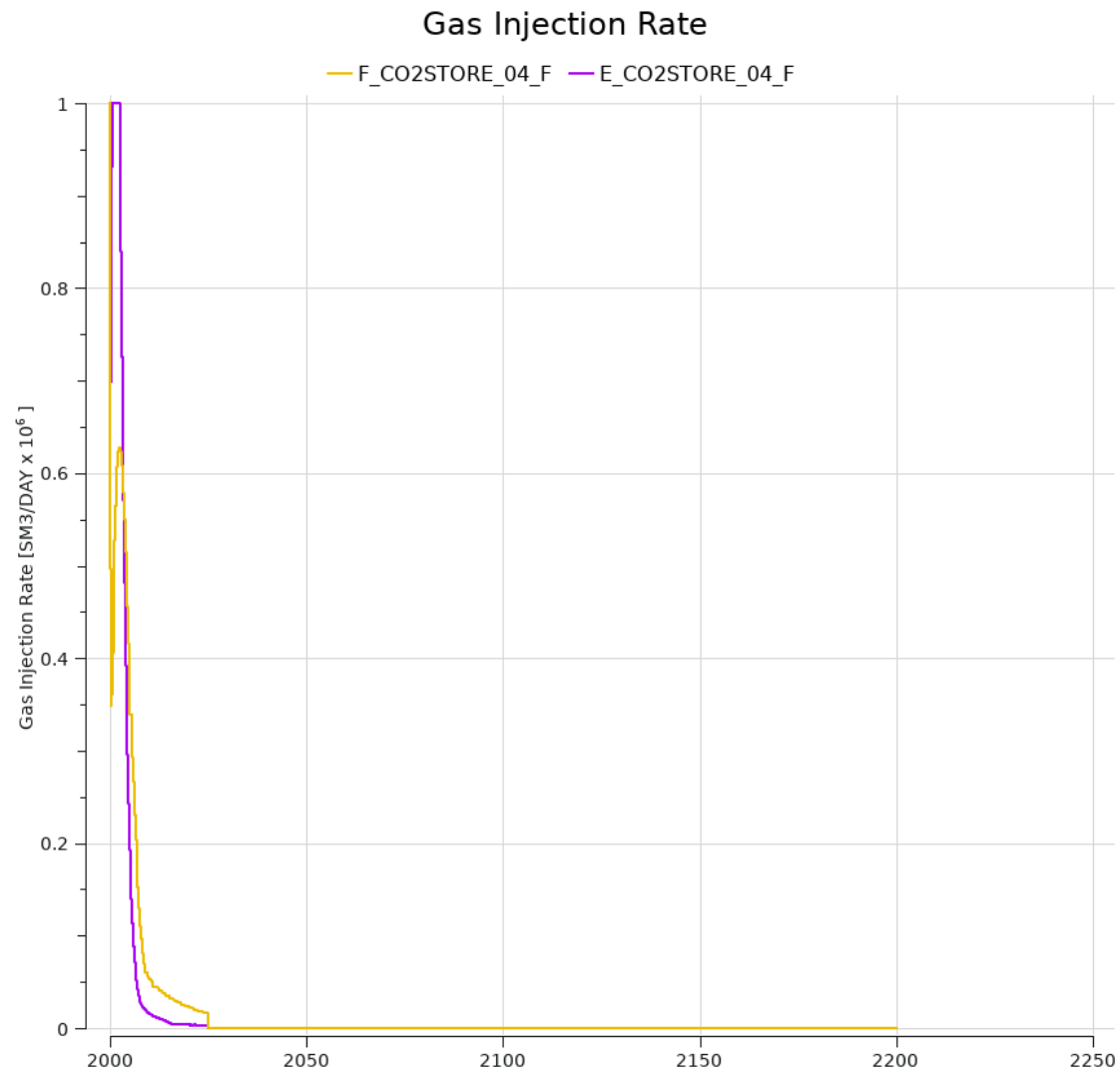
- Model with salinity

Pressure average value

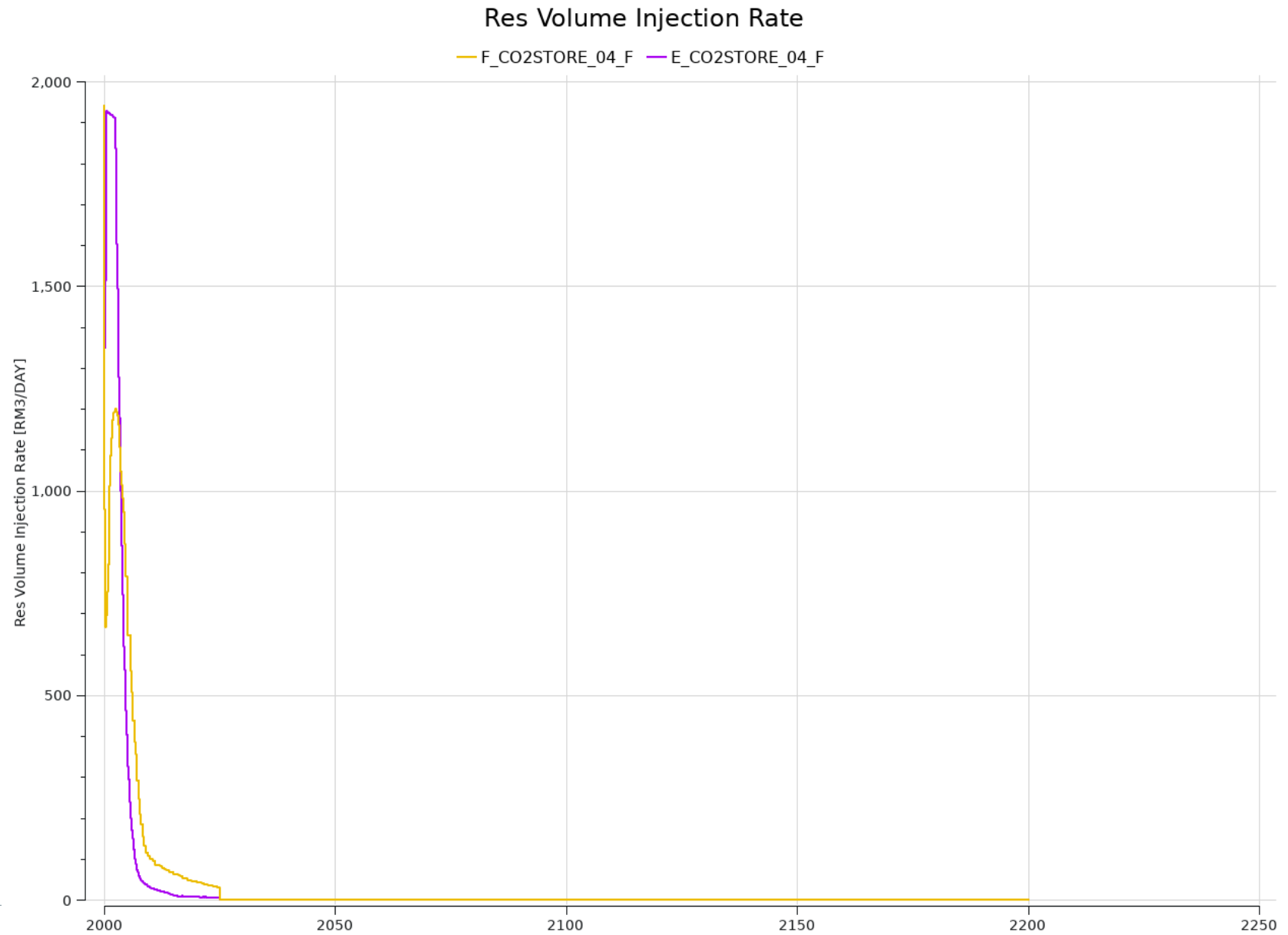
Average pressure



Injection rates

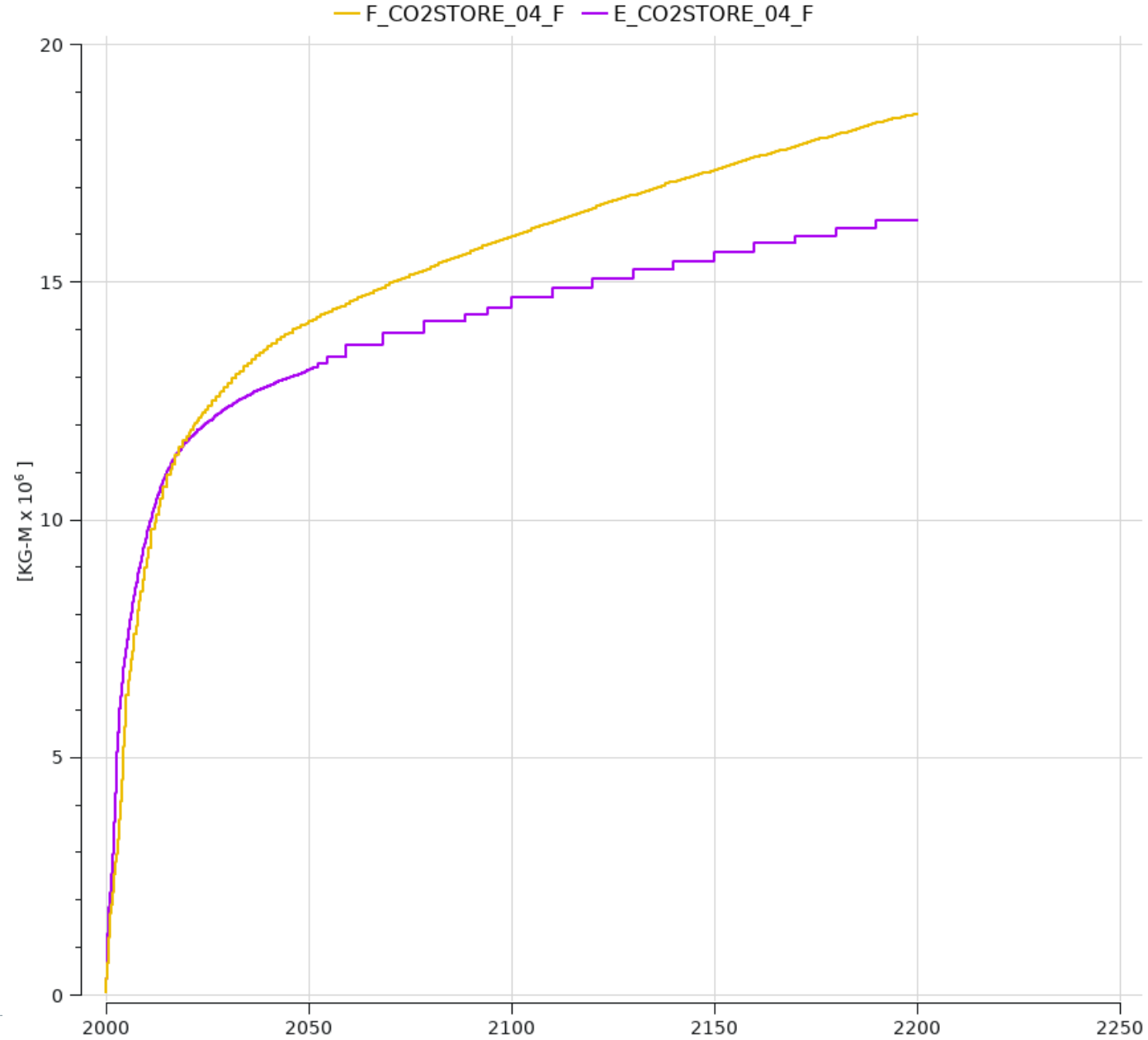


Reservoir rate




FWCD

Dissolution



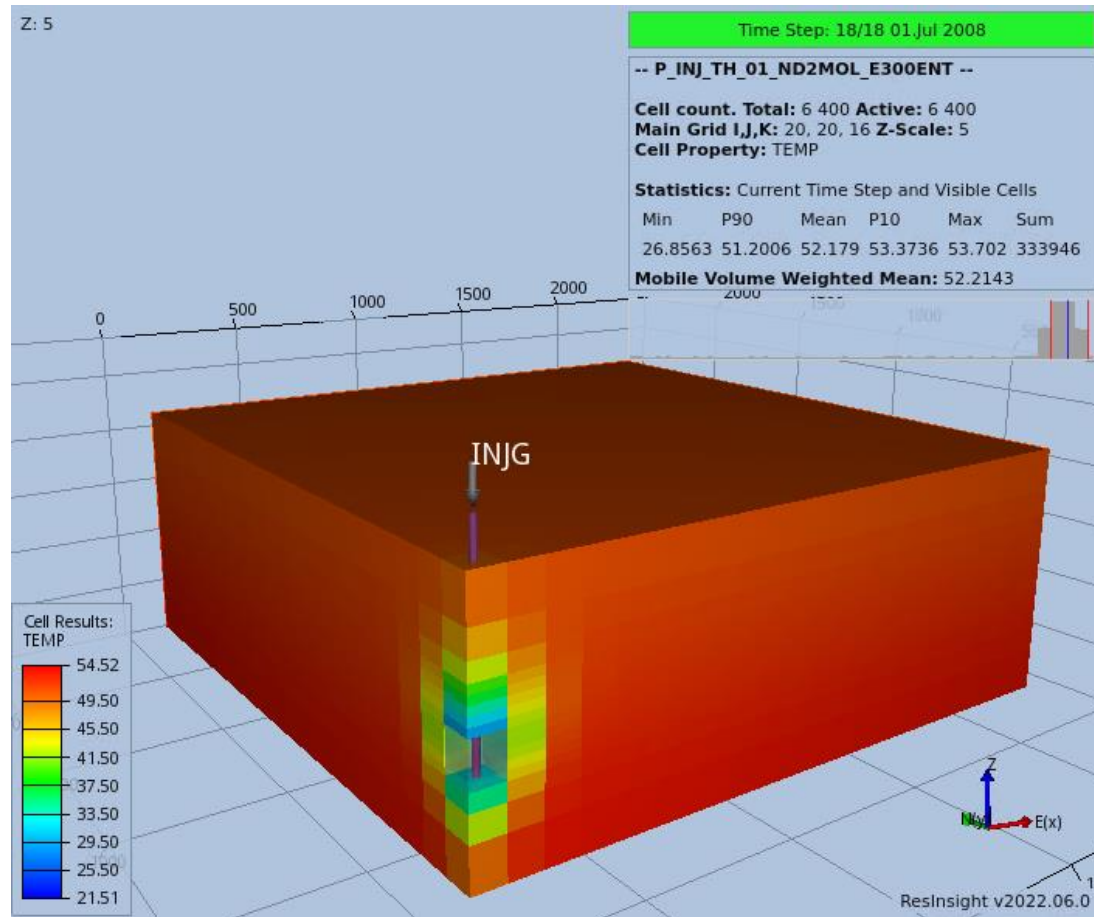
Observations

- For models without salinity
 - Generally good agreement between simulators
- For models with salinity deviations are observed
 - For FWCD the mixing model matters
 - For reservoir rates we need to understand difference

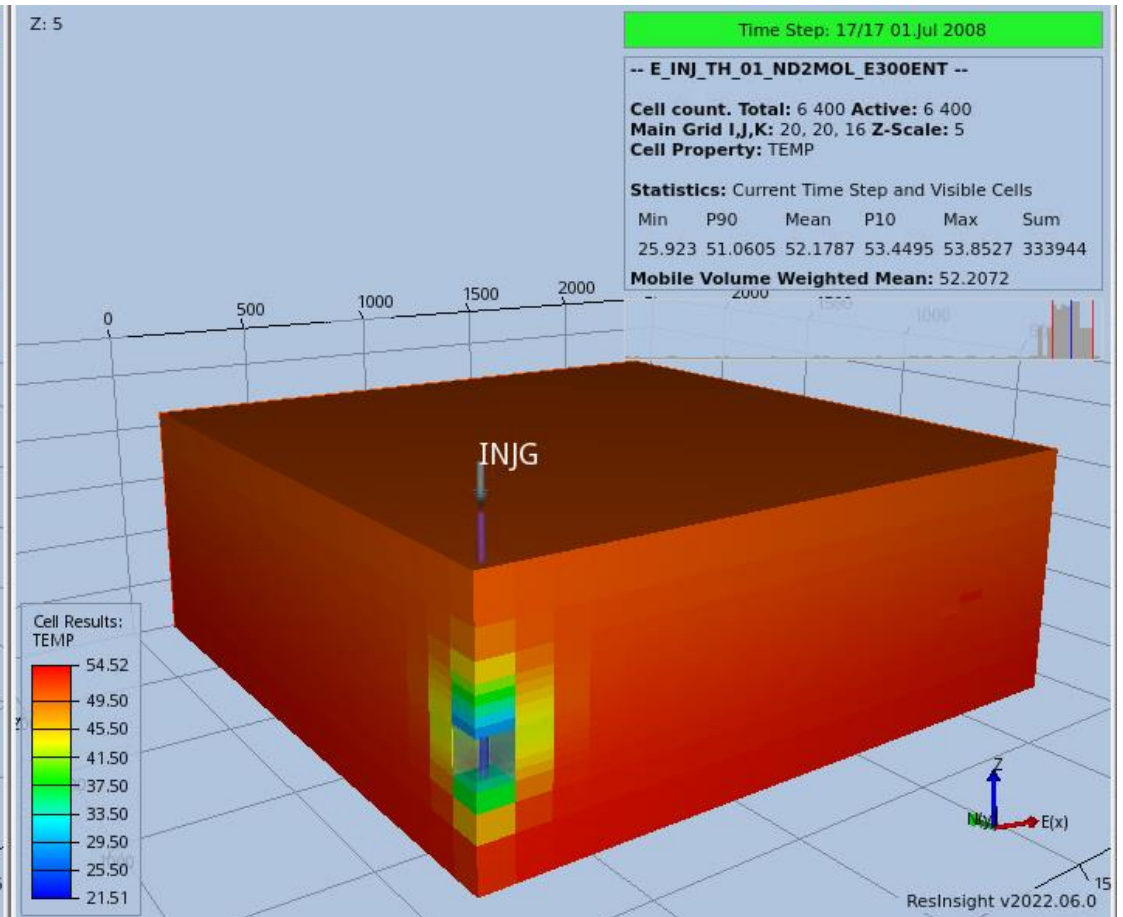
- Thermal simulations 
 - With salinity

The grids

Pflotran

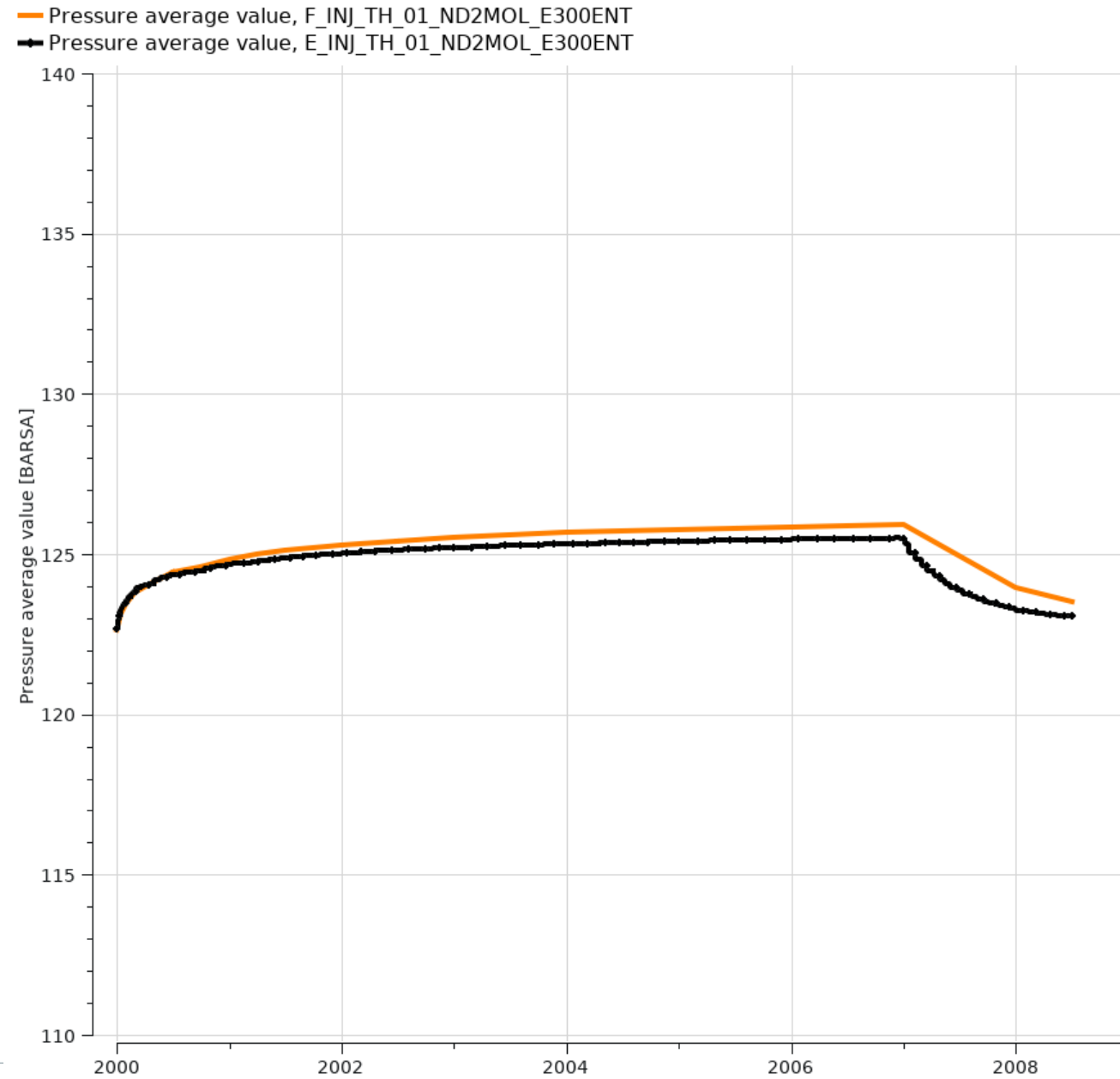


E300

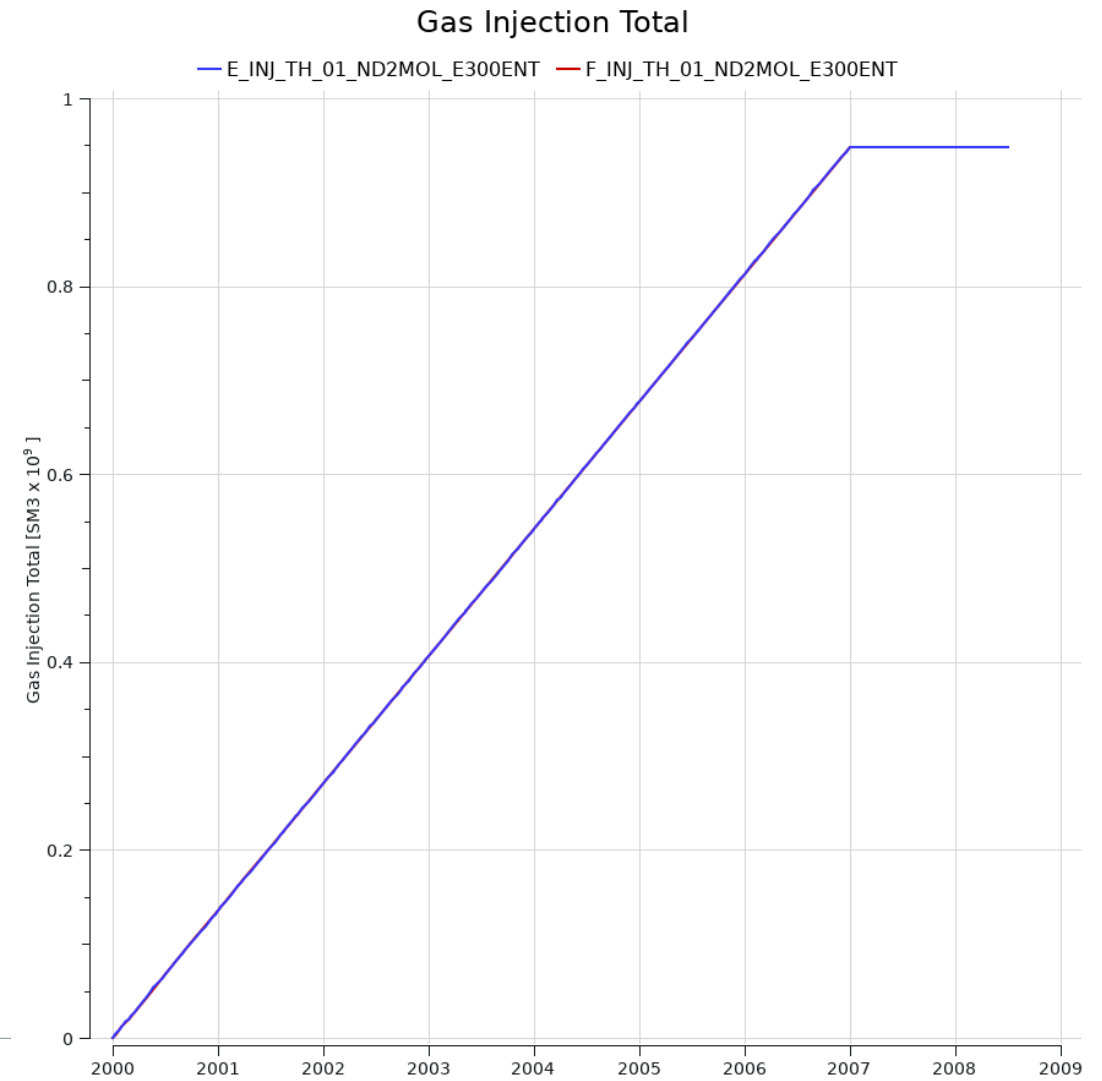
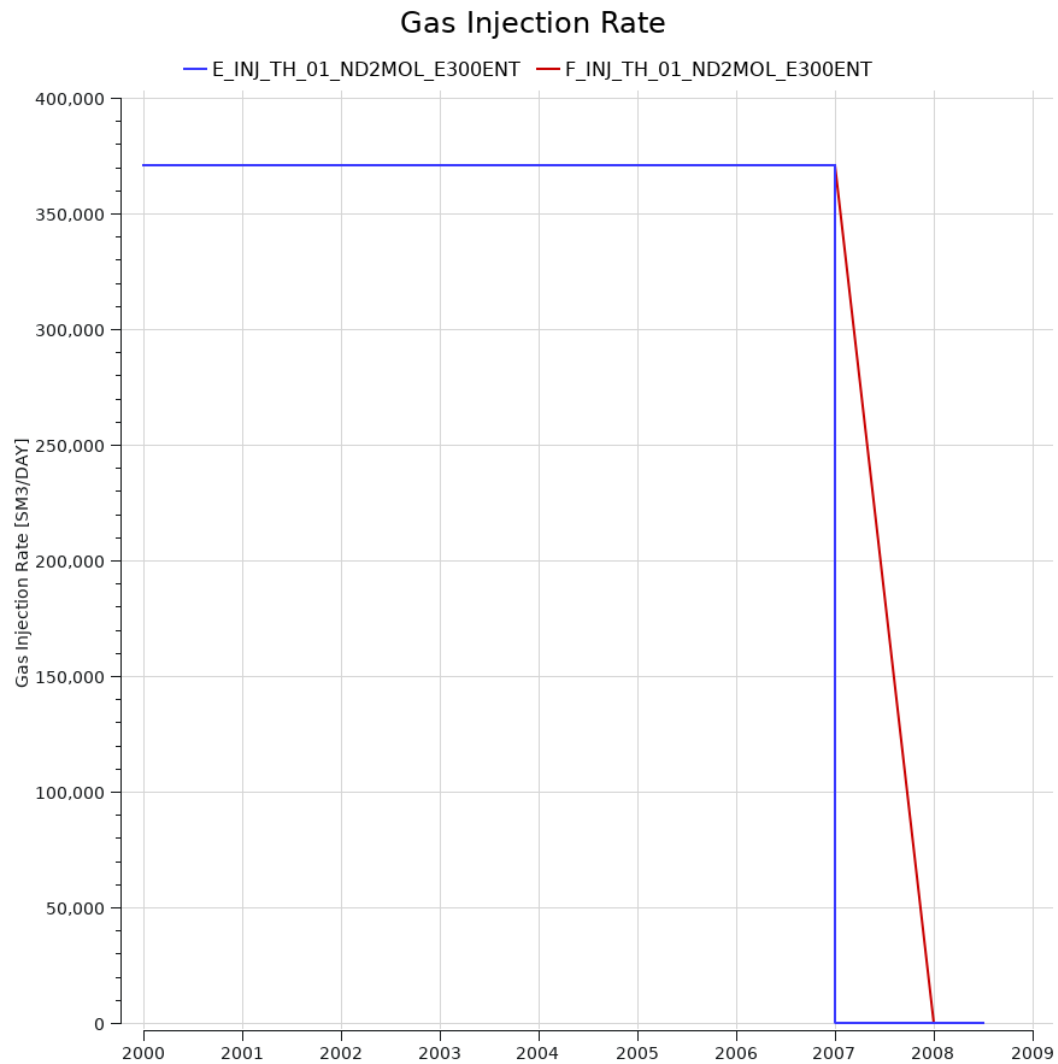


Plot 6

Average pressure

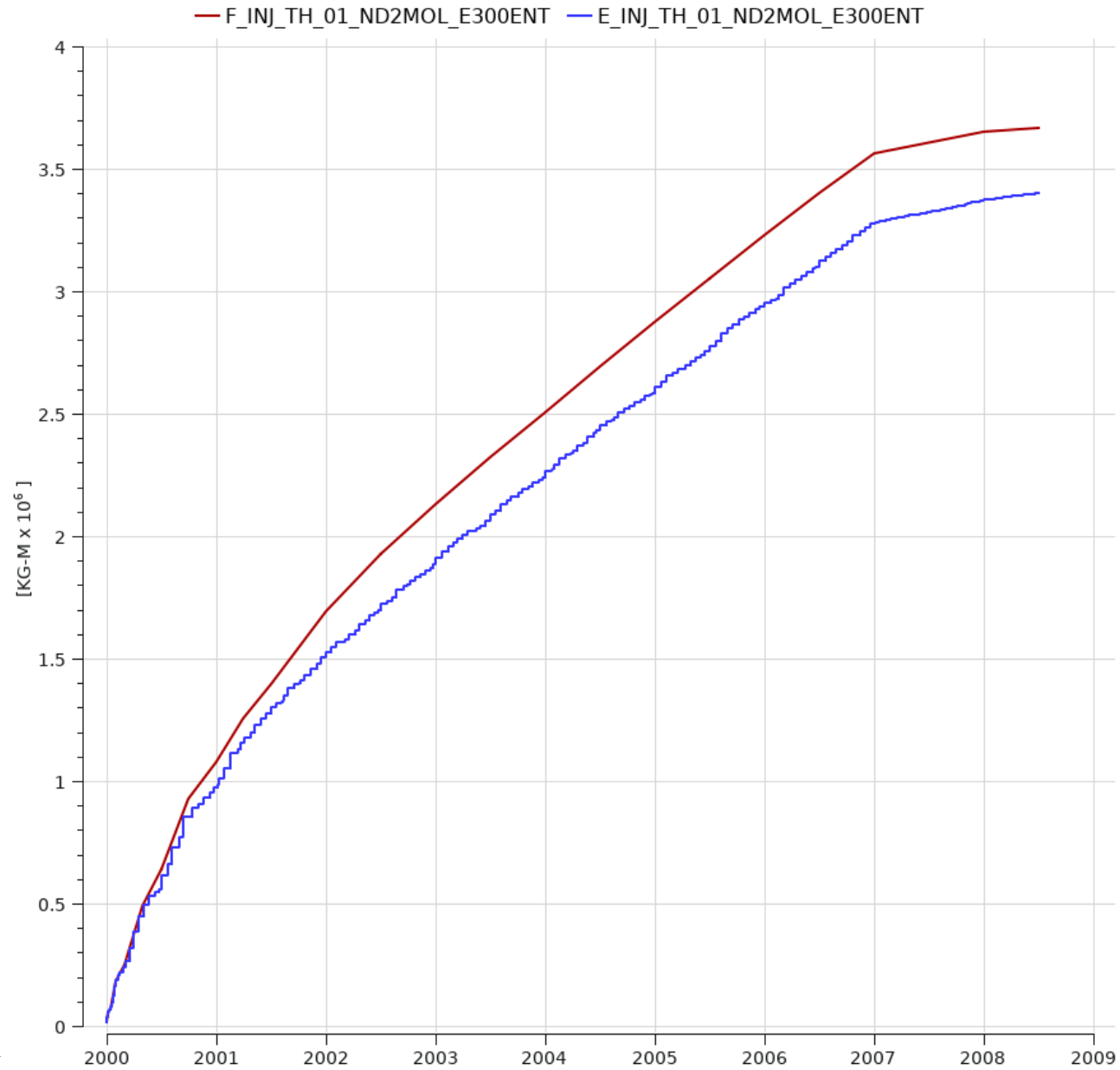


Injection rates



FWCD

Dissolution



CO2 densities

$$P = \left(\frac{RT_K}{V - b_{mix}} \right) - \left(\frac{a_{mix}}{T_K^{1/2} V(V + b_{mix})} \right)$$

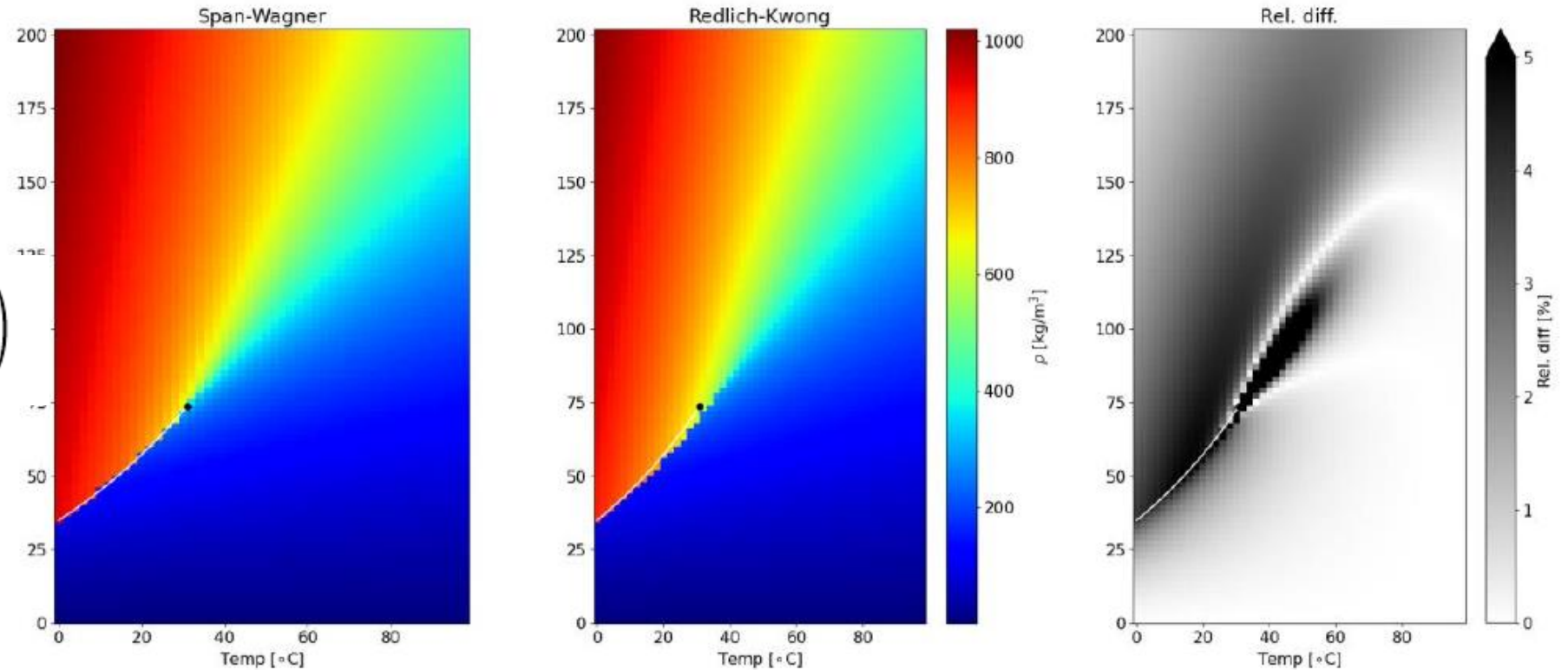


Figure 3. Comparison of density between Span-Wagner used by OPM Flow and Redlich-Kwong.

Solubility model and ACTCO2S

The mutual solubilities of water and CO₂ are then expressed as follows:

$$y_{\text{H}_2\text{O}} = A(1 - x_{\text{CO}_2})$$

$$x_{\text{CO}_2} = B(1 - y_{\text{H}_2\text{O}})$$

with parameters A and B defined as (using values of K given by Eqs. 5–7)

$$A = \frac{K_{\text{H}_2\text{O}} \gamma_{\text{H}_2\text{O}}}{\Phi_{\text{H}_2\text{O}} P_{\text{tot}}}$$

$$B = \frac{\Phi_{\text{CO}_2} P_{\text{tot}}}{55.508 \gamma_{\text{CO}_2} K_{\text{CO}_2}}$$

$$\ln(\gamma_{\text{H}_2\text{O}}) = (A_M - 2A_M x_{\text{H}_2\text{O}}) x_{\text{CO}_2}^2 \quad (12)$$

$$\ln(\gamma_{\text{CO}_2}) = 2A_M x_{\text{CO}_2} x_{\text{H}_2\text{O}}^2 \quad (13)$$

In these equations, A_M is a Margules parameter that, after several tests, we chose to express as a function of temperature, as follows:

$$A_M = 0 \text{ (thus } \gamma_{\text{CO}_2} \text{ and } \gamma_{\text{H}_2\text{O}} = 1) \text{ at } T \leq 100^\circ\text{C} \quad (14)$$

$$A_M = a(T_K - 373.15) + b(T_K - 373.15)^2 \text{ at } T > 100^\circ\text{C} \quad (15)$$

Duan and Sun 2003 (Spycher and Pruess 2005)

Rumpf et al. 1994 (Spycher and Pruess 2005)

Spycher and Pruess 2009

The effect of salt

$$\gamma'_{\text{CO}_2} = \left(1 + \frac{\sum m_{i \neq \text{CO}_2}}{55.508} \right) \exp\{2\lambda(m_{\text{Na}} + m_{\text{K}} + 2m_{\text{Ca}} + 2m_{\text{Mg}}) + \xi m_{\text{Cl}}(m_{\text{Na}} + m_{\text{K}} + m_{\text{Ca}} + m_{\text{Mg}}) - 0.07 m_{\text{SO}_4}\}$$

$$B' = \frac{\Phi_{\text{CO}_2} P_{\text{tot}}}{55.508 \gamma_{\text{CO}_2} \gamma'_{\text{CO}_2} K_{\text{CO}_2}}$$

The mutual solubilities of water and CO₂ are then expressed as follows:

$$y_{\text{H}_2\text{O}} = A(1 - x_{\text{CO}_2})$$

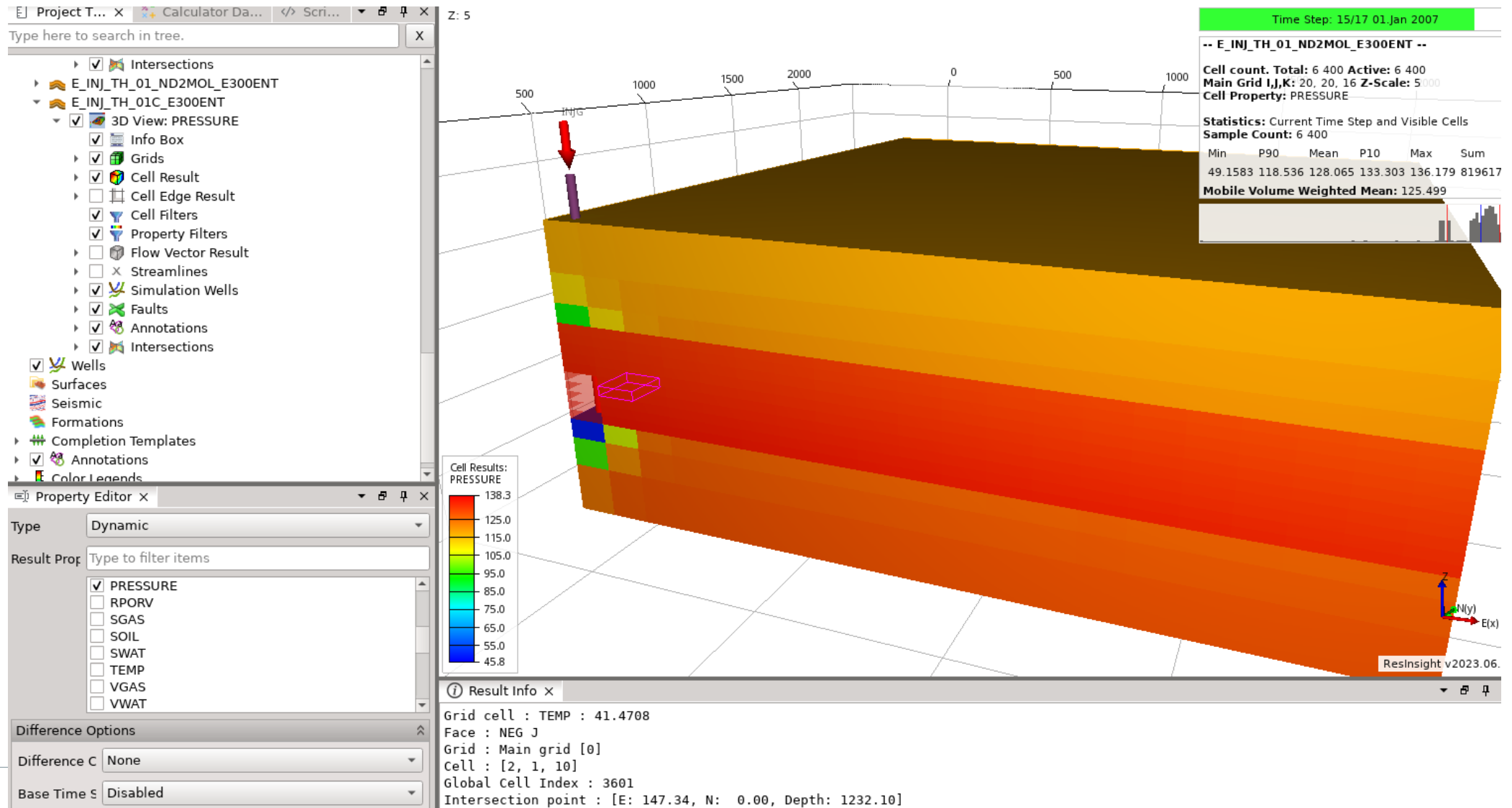
$$x_{\text{CO}_2} = B(1 - y_{\text{H}_2\text{O}})$$

with parameters A and B defined as (using values of K given by Eqs. 5–7)

$$A = \frac{K_{\text{H}_2\text{O}} \gamma_{\text{H}_2\text{O}}}{\Phi_{\text{H}_2\text{O}} P_{\text{tot}}}$$

$$B = \frac{\Phi_{\text{CO}_2} P_{\text{tot}}}{55.508 \gamma_{\text{CO}_2} K_{\text{CO}_2}}$$

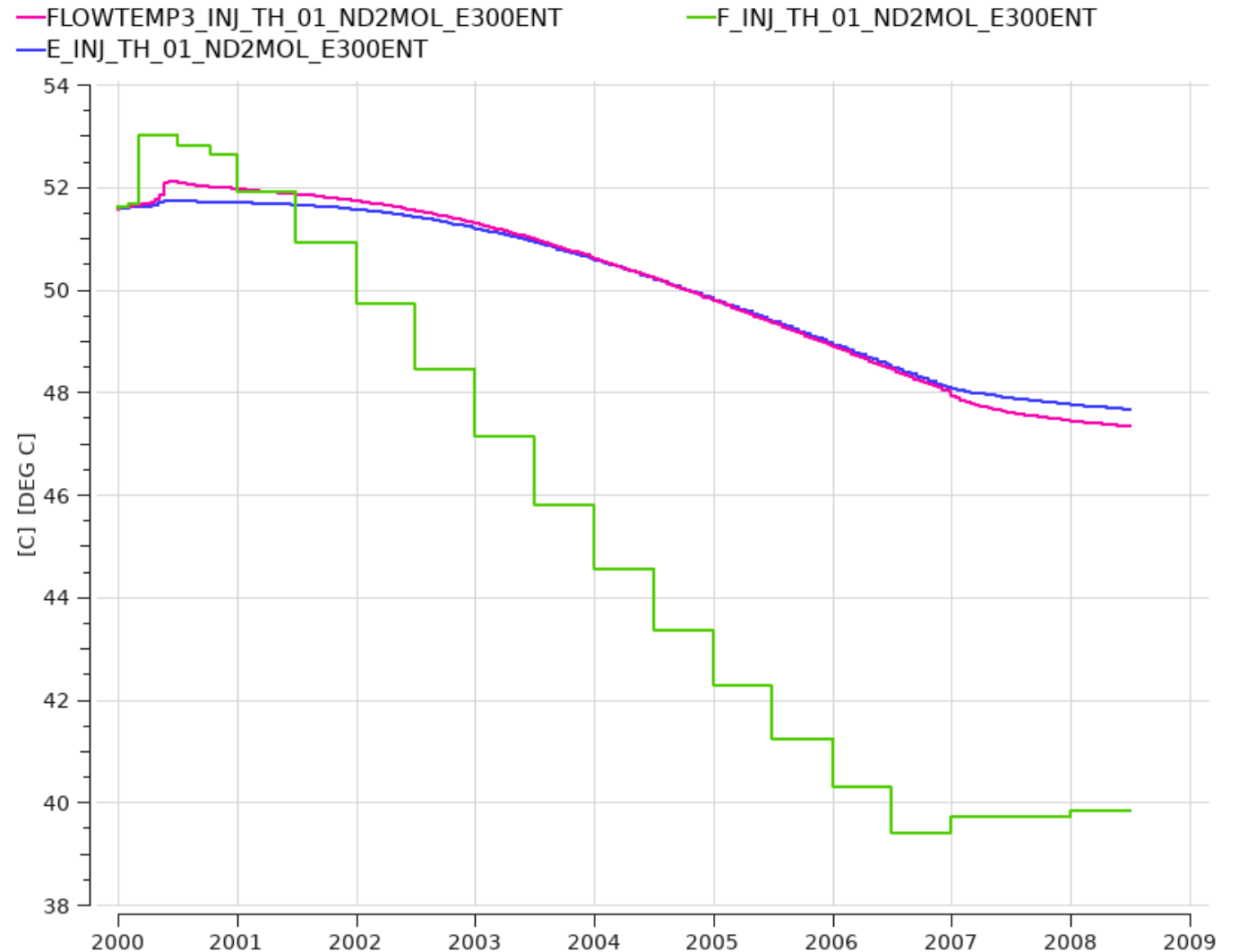
The models



Block : 2,1,4, BTEMP

Where we left off

- Deviation between E300/pFlotran and Flow was close to 9 degrees Celsius
- Pflotran was claimed to use Span-Wagner?
- It was commented that temperature could be critical, Sleipner as example



Span & Wagner

```
← → ↻ 🏠 🔒 raw.githubusercontent.com/OPM/opm-common/master/src/opm/material/components/co2tables.inc 📄 ☆ 🔴 📺 🧩 📱 📄
```

```
/* Tables for CO2 fluid properties calculated according to Span and
 * Wagner (1996).
 *
 * THIS AN AUTO-GENERATED FILE! DO NOT EDIT IT!
 *
 * Temperature range: 280.000 K to 400.000 K, using 200 sampling points
 * Pressure range: 0.100 MPa to 100.000 MPa, using 500 sampling points
 *
 * Generated using:
 *
 * ./extractproperties 280.0 400.0 200 1e5 100e6 500
 */
struct TabulatedDensityTraits {
    typedef double Scalar;
    static const char *name;
    static const int    numX = 200;
    static const Scalar xMin;
    static const Scalar xMax;
    static const int    numY = 500;
    static const Scalar yMin;
    static const Scalar yMax;

    static const std::vector<std::vector<Scalar>> vals;
};

inline const double TabulatedDensityTraits::xMin = 2.8000000000000000e+02;
inline const double TabulatedDensityTraits::xMax = 4.0000000000000000e+02;
inline const double TabulatedDensityTraits::yMin = 1.0000000000000000e+05;
inline const double TabulatedDensityTraits::yMax = 1.0000000000000000e+08;
inline const char *TabulatedDensityTraits::name = "density";

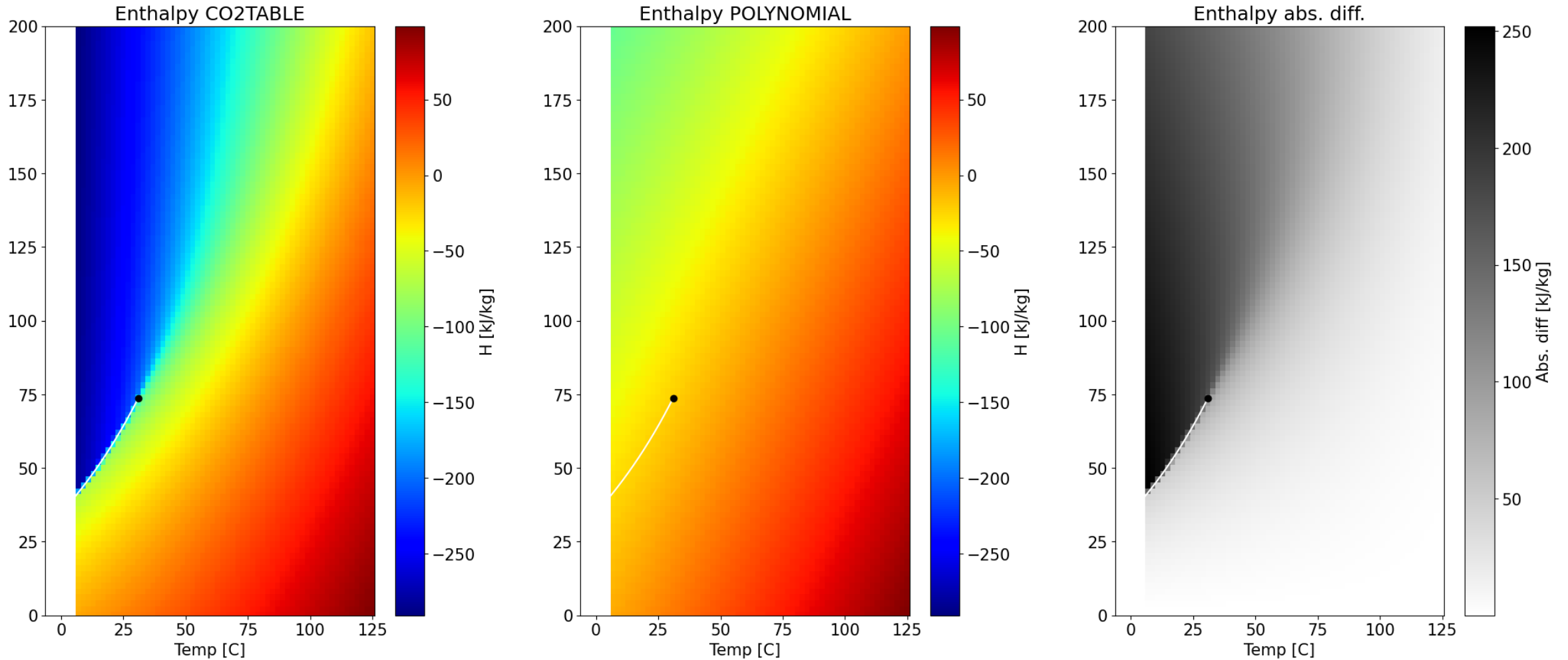
inline const std::vector<std::vector<double>> TabulatedDensityTraits::vals =
{
    {
        1.902062465274410e+00,    5.782408947482105e+00,    9.764909779046345e+00,    1.385694970929571e+01,    1.806682704790397e+01,
        2.240391789044656e+01,    2.687888030157356e+01,    3.150391104248634e+01,    3.629307240336141e+01,    4.126271410164965e+01,
        4.643202645965111e+01,    5.182377862839375e+01,    5.746532380925083e+01,    6.339000019560863e+01,    6.963913698776773e+01,
        7.626502018649226e+01,    8.333544920754456e+01,    9.094107565501655e+01,    9.920794710517262e+01,    1.083206802964683e+02,
        1.185701068818210e+02,    8.854494190604881e+02,    8.879785335109976e+02,    8.904262882902154e+02,    8.927992468133788e+02,
        8.951031318186318e+02,    8.973429696398650e+02,    8.995232040155760e+02,    9.016477869623172e+02,    9.037202521331737e+02,
        9.057437746274870e+02,    9.077212201969121e+02,    9.096551860638534e+02,    9.115480350401762e+02,    9.134019242461904e+02,
        9.152188294415341e+02,    9.170005657625713e+02,    9.187488054960497e+02,    9.204650933921371e+02,    9.221508599218751e+02,
        9.238074328074559e+02,    9.254360470933484e+02,    9.270378539784197e+02,    9.286139285909143e+02,    9.301652768573648e+02,
        9.316928415915886e+02,    9.331975079096469e+02,    9.346801080600167e+02,    9.361414257445779e+02,    9.375821999946785e+02,
        9.390031286571951e+02,    9.404048715376014e+02,    9.417880532405302e+02,    9.431532657427566e+02,    9.445010707288508e+02,
        9.458320017157859e+02,    9.471465659893925e+02,    9.484452463727065e+02,    9.497285028437043e+02,    9.509967740178902e+02,
        9.522504785092801e+02,    9.534900161817900e+02,    9.547157693016472e+02,    9.559281036002363e+02,    9.571273692557639e+02,
        9.583139018012021e+02,    9.594880229651809e+02,    9.606500414517863e+02,    9.618002536646125e+02,    9.629389443798724e+02,
        9.640663873728772e+02,    9.651828460017828e+02,    9.662885737521152e+02,    9.673838147452494e+02,    9.684688042137223e+02,
```

CO2 Enthalpy main difference

```

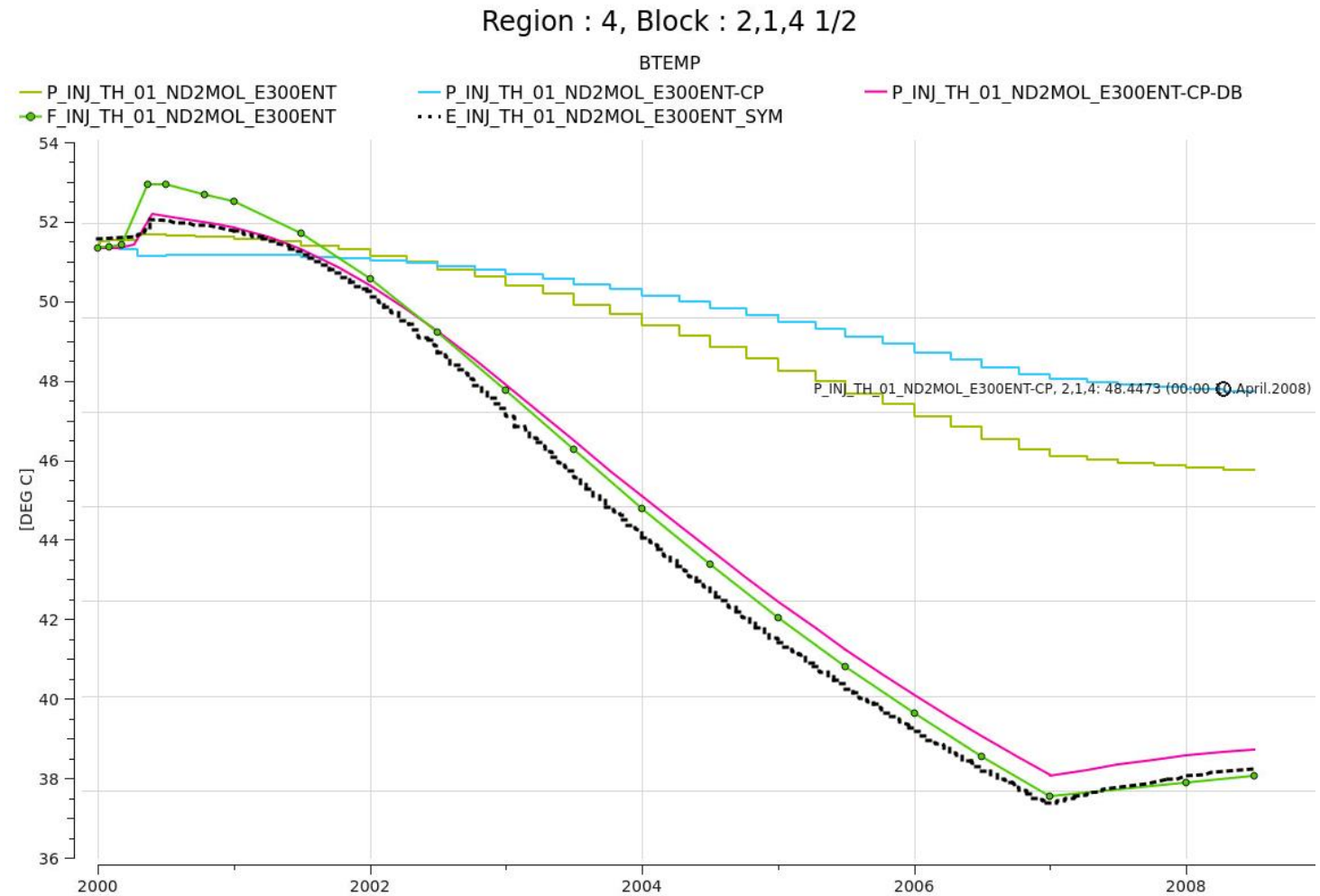
 5  opm/material/components/CO2.hpp
  @@ -169,7 +169,10 @@ class CO2 : public Component<Scalar, CO2<Scalar>>
 169 169         const Evaluation& pressure,
 170 170         bool extrapolate = false)
 171 171     {
 172 -     return tabulatedEnthalpy.eval(temperature, pressure, extrapolate);
 172 +     // TEST 2nd degree polynomial fitted with Coolprop data in temperature
 173 +     // range (273.15, 403.15) with reference state T=288.15 K (=15 C) and p = 101325 Pa
 174 +     return (temperature - 273.15 - 15)*(8.42323594e+02 + 4.54513769e-01*(temperature - 273.15 - 15)) - 0.005 * (pressure - 1.01325e5);
 175 +     // return tabulatedEnthalpy.eval(temperature, pressure, extrapolate);
 173 176     }
 174 177
 175 178     /*!
  
```

Comparing Enthalpy representations



Comparison with Pflotran and E300

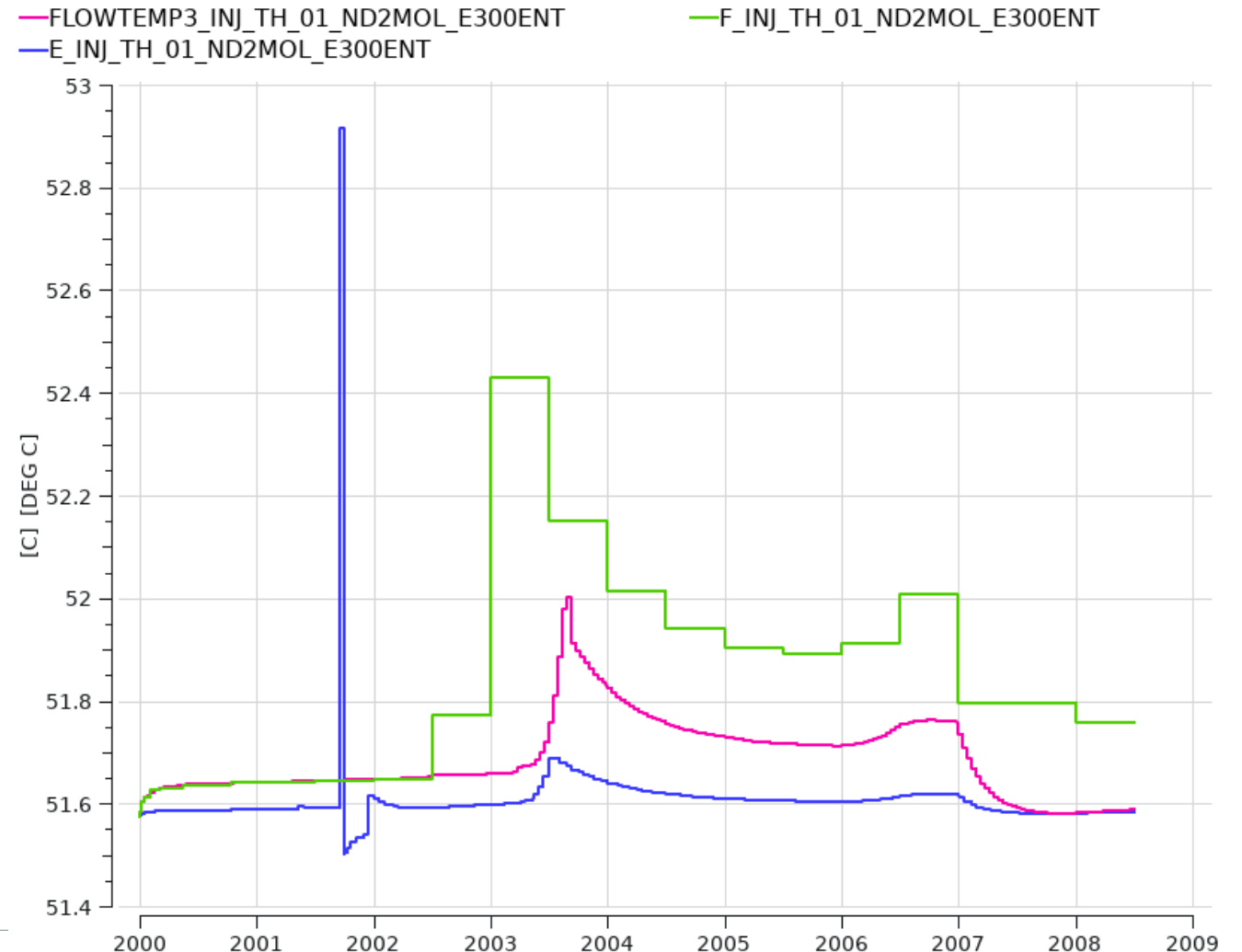
- Pflotran confirmed not to have heat of dissolution
- Both Pflotran and E300 confirms the low temperatures of OPM Flow when Span-Wagner properties for CO2 enthalpy are used



CO2 dissolution secondary effect

- Dissolution effect for CO2 gives a temperatur «bump»
- Removing dissolution model from Flow removes the main difference
- What about temperature effect of vaporization?

Block : 7,7,4, BTEMP



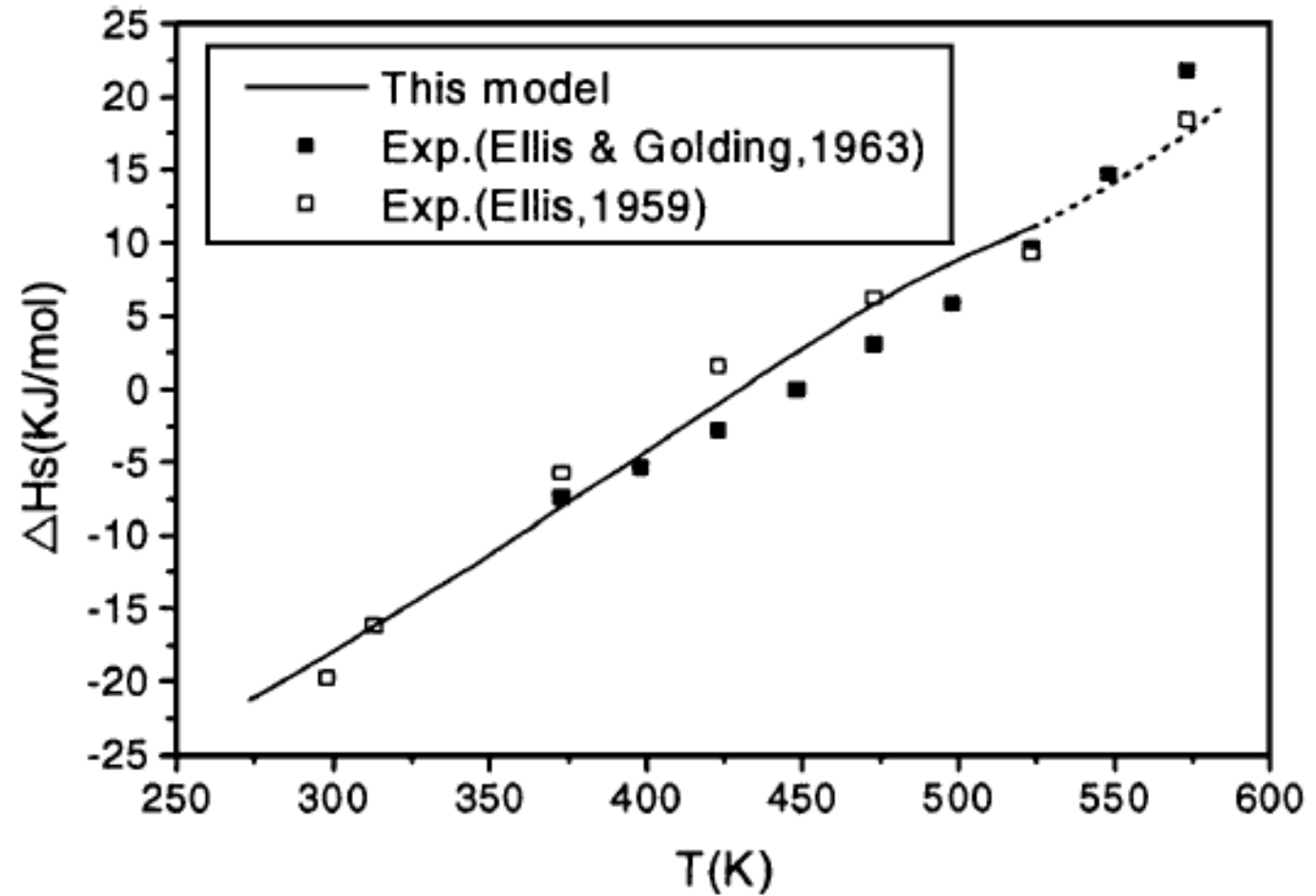


Fig. 6. The heat of solution of CO₂ in water (the model of this study vs. experimental data).

CO2 Dissolution in Flow

```

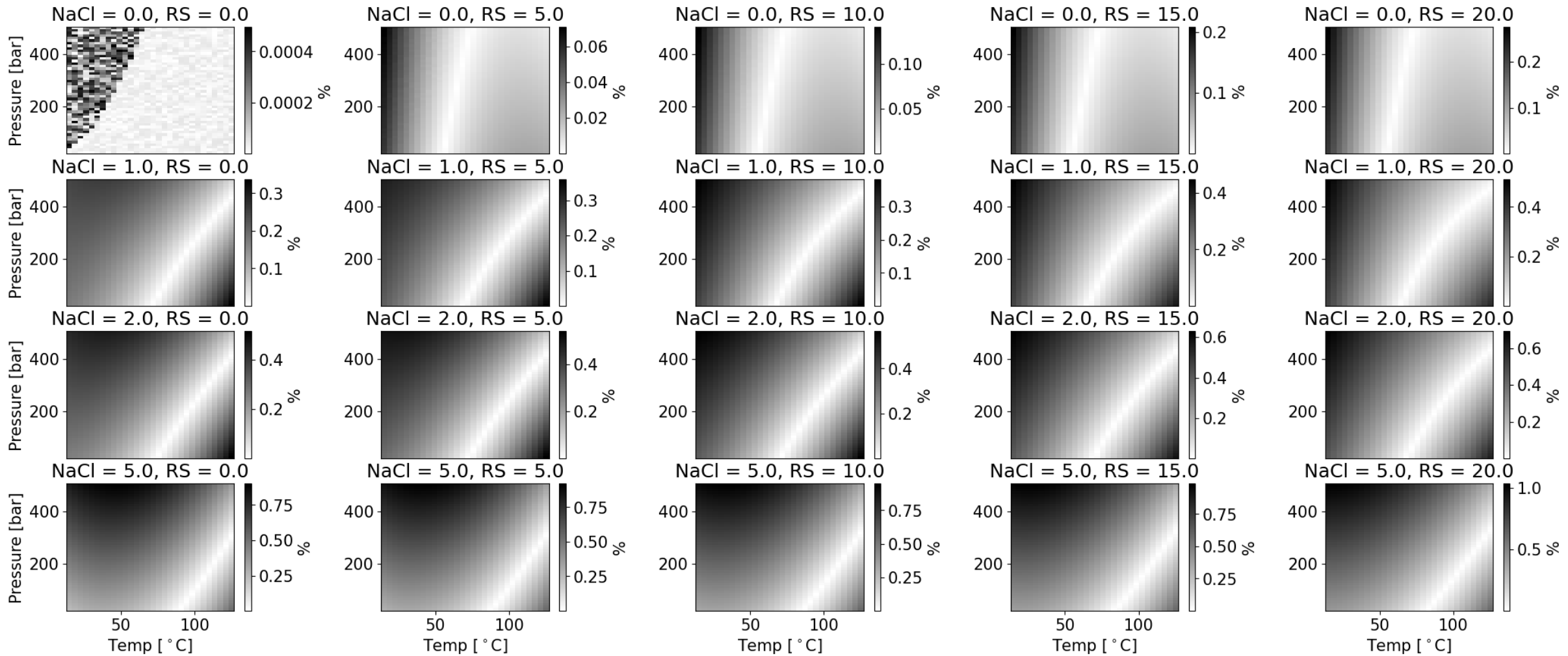
opm/material/fluidsystems/blackoilpvt/BrineCo2Pvt.hpp
Viewed

@@ -612,10 +612,10 @@ class BrineCo2Pvt
612 612      /* heat of dissolution for CO2 according to Fig. 6 in Duan and Sun 2003. (kJ/kg)
613 613          In the relevant temperature ranges CO2 dissolution is
614 614          exothermal */
615 -      delta_hCO2 = (-57.4375 + T * 0.1325) * 1000/44;
615 +      // delta_hCO2 = (-57.4375 + T * 0.1325) * 1000/44;
616 616
617 617      /* enthalpy contribution of CO2 (kJ/kg) */
618 -      hg = CO2::gasEnthalpy(T, p, extrapolate)/1E3 + delta_hCO2;
618 +      hg = CO2::gasEnthalpy(T, p, extrapolate)/1E3; //+ delta_hCO2;
619 619
620 620      /* Enthalpy of brine with dissolved CO2 */
621 621      return (h_ls1 - X_CO2_w*hw + hg*X_CO2_w)*1E3; /*J/kg*/

```

Water density (diff with Ezrkohi, RS is CO2 content)

REL_DIFF



Density models for saline water in E300, Intersect, Pflotran and Flow

- Flow uses Garcia for CO₂ solution impact on water density and Batzle & Wang salinity impact on density of water, ref. <https://library.seg.org/doi/10.1190/1.1443207> <https://www.osti.gov/biblio/790022>
- E300 and Intersect uses Ezrokhi for both (CO₂ solution and salinity impact on water density). Model described in book: Zaytsev, I.D., Aseyev, G.G.: Properties of Aqueous Solutions of Electrolytes 1993.
- Pflotran supports Batzle & Wang in addition to Ezrokhi for density of saline water, while the CO₂ solution impact is from Duan & Sun, ref. https://docs.opengosim.com/manual/input_deck/thermodynamic_props/eos_water/
- Note that Batzle & Wang is more accurate than Ezrokhi. However, as implemented in Flow it only takes NaCl into account, will need to be extended to take other salts into account. While Ezrokhi allows to manually provide parameters to the model to account for salt combination.